## **Chapter 3. The Quantum Mechanical Postulates**

**Postulates 1:** A quantum state is completely specified by a <u>wave</u> <u>function</u>, which is the solution of the Schrödinger equation.

<u>Born's interpretation</u>: The probability of finding the system between x and x+dx is proportional to  $|\Psi(x)|^2 dx$ .

As a result, a wavefunction must be

- i. <u>normalized</u>,  $\int |\Psi|^2 dx = \int \Psi^* \Psi dx = 1$
- ii. single valued,
- iii. finite,
- iv. continuous in zeroth and first derivatives.

Generalization:

3D: 
$$\Psi^*(x, y, z)\Psi(x, y, z)dxdydz$$
  
2 particles:  $\Psi^*(\mathbf{r}_1, \mathbf{r}_2)\Psi(\mathbf{r}_1, \mathbf{r}_2)d\mathbf{r}_1d\mathbf{r}_2$ , where  $\mathbf{r} = (x, y, z)$ 

 $\Psi$  is the probability amplitude, and  $|\Psi|^2$  is the probability density.

**Postulate 2:** Every observable has a corresponding quantum operator. Measuring an observable is equivalent to applying the operator on a wave function.

All quantum operators are <u>Hermitian</u>

$$\int \psi^* [\hat{O}\varphi] dx = \int [\hat{O}\psi]^* \varphi dx$$

whose eigenvalues are always real and eigenfunctions are <u>complete</u> and <u>orthonormal</u>:

$$\int \Psi_n^* \Psi_m dx = \delta_{nm}$$

where the Kroenecker delta is defined as

$$\delta_{nm} = \begin{cases} 1, & \text{if } n = m \\ 0, & \text{if } n \neq m \end{cases}$$

**Postulate 3**: Each measurement only yields an <u>eigenvalue as the</u> <u>observable</u>.

**Postulate 4**: For many measurements, the average observable is given by the <u>expectation value</u>:

$$\langle a \rangle = \frac{\int \Psi^* \hat{A} \Psi dx}{\int \Psi^* \Psi dx}$$

If  $\Psi$  is an eigenstate  $(\Psi = \Psi_n, \hat{A}\Psi_n = a_n\Psi_n)$ , then  $\langle a \rangle$  is simply  $a_n$ :

$$\langle a \rangle = \int \Psi^* \hat{A} \Psi dx = \int \Psi_n^* a_n \Psi_n dx = a_n \int \Psi_n^* \Psi_n dx = a_n$$

what if  $\Psi$  is not an eigenstate?

$$\Psi(x) = \sum_{n} b_n \Psi_n \qquad \text{(completeness)}$$

Superposition principle

$$\langle a \rangle = \int \left( \sum_{n} b_{n} \Psi_{n} \right)^{*} \hat{A} \left( \sum_{m} b_{m} \Psi_{m} \right) d\tau$$

$$= \int \left( \sum_{n} b_{n} \Psi_{n} \right)^{*} \left( \sum_{m} b_{m} a_{m} \Psi_{m} \right) d\tau \qquad (\int \Psi_{n}^{*} \Psi_{m} d\tau = \delta_{nm})$$

$$= \sum_{n} \sum_{m} b_{n}^{*} b_{m} a_{m} \int \Psi_{n}^{*} \Psi_{m} d\tau$$

$$= \sum_{n} |b_{n}|^{2} a_{n}$$

- i. Each measurement always yields one of the eigenvalues,
- ii. The chance for a particular eigenvalue  $(a_n)$  to appear in a measurement is given by  $|b_n|^2$ .
- iii. Average value of many measurements is given by  $\langle a \rangle$ .

Interestingly, the precise state of the system is unknown until the measurement, which collapses the system to an eigenstate.

Example: an electron in a molecule (e.g., H<sub>2</sub>) is described by a molecular orbital, which is a superposition of atomic orbitals:

$$\Phi^{MO} = \sum_{n} c_n \Psi_n^{AO} = c_1 1 s_1 + c_2 1 s_2$$

Same as double slit.